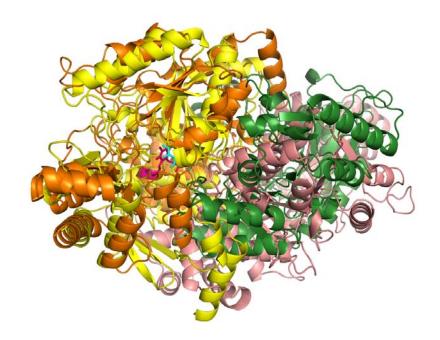
# The Structure and Function of Biomolecules



#### The student is expected to:

TEKS 9A compare the structures and functions of different types of biomolecules, including carbohydrates, lipids, proteins, and nucleic acids



#### **KEY CONCEPT**

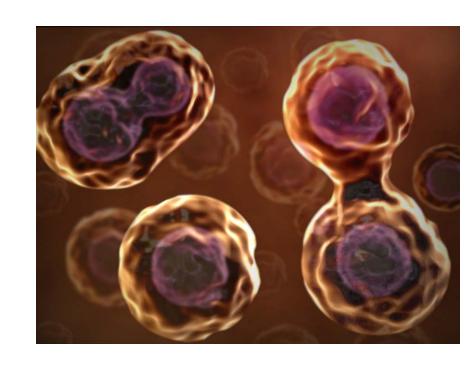
Carbon-based molecules are the foundation of life.



## **Humans use food for:**

1. Energy

2. Growth of new cells and the repair of existing cells, tissues, organs, etc.



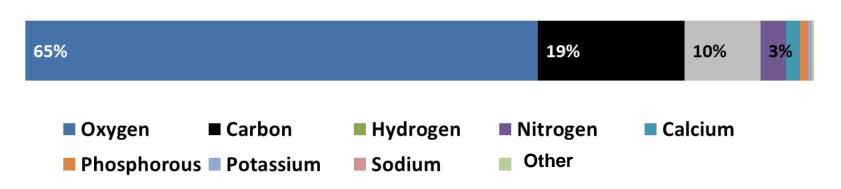
## What is Food made up of?

Six chemical elements: C, H, O, N, P, S

Salts Na, Mg, Cl, K, Ca

Three trace elements: Fe, Cu, Zn

% of elements in the human body



## Carbon atoms have unique bonding properties.

- Carbon forms covalent bonds with up to four other atoms, including other carbon atoms.
- Carbon-based molecules have three general types of structures.
  - straight chain
  - branched chain
  - ring

Straight chain

A simplified structure can also be shown as:

$$CH_3$$
— $CH_2$ — $CH_2$ — $CH$ = $CH_2$ 

Pentene

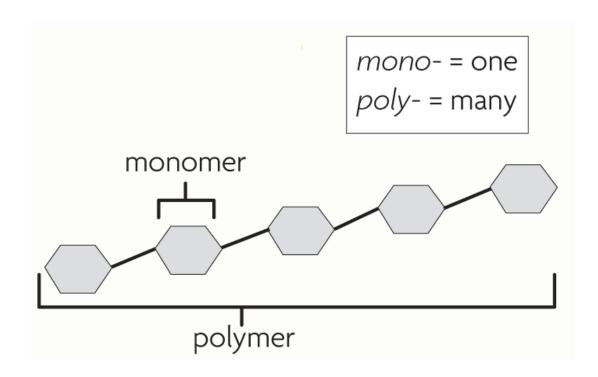
Branched chain

Hexane

Ring
O
CH
CH
CH
CH
OH
Vanillin

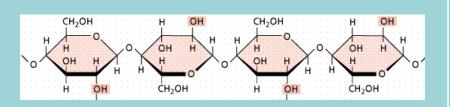
## Many carbon-based molecules are made of many small subunits bonded together.

- Monomers are the individual subunits.
- Polymers are made of many monomers.

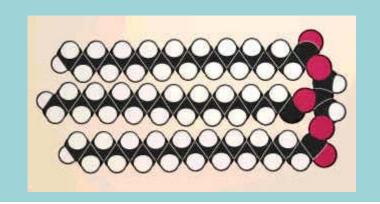


## Four main types of carbon-based molecules are found in living things.

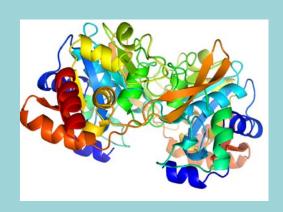
## 1. Carbohydrates



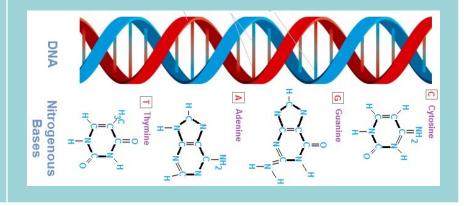
## 2. Lipids



### 3. Proteins



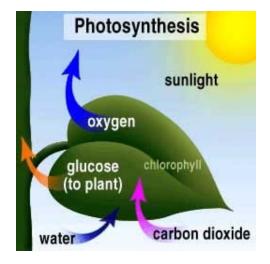
## 4. Nucleic Acids



## **Types of Carbohydrates**

- Carbohydrates include sugars and starches.
- Monosaccharides are simple sugars.
- Polysaccharides include starches, cellulose, and glycogen.

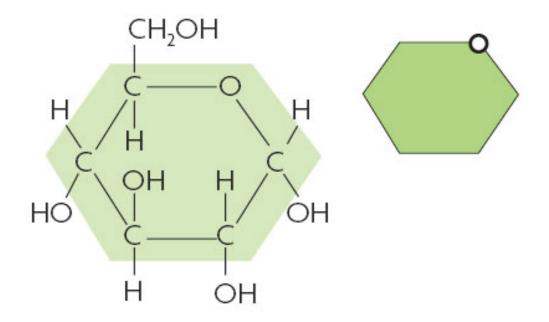




Glucose is a carbohydrate!

## **Structure of Carbohydrates**

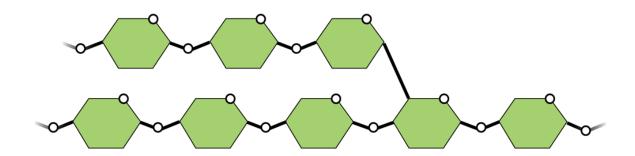
Carbohydrates are made of carbon, hydrogen, and oxygen.



Glucose ( $C_6H_{12}O_6$ ) can be ring shaped and is often shown as a simplified hexagon.

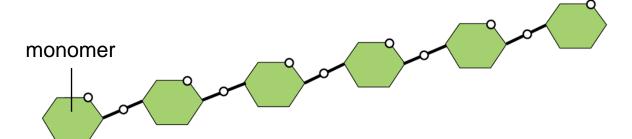
## **Structure of Carbohydrates**

#### Polymer (starch)



Starch is a polymer of glucose monomers that often has a branched structure.

#### Polymer (cellulose)



Cellulose is a polymer of glucose monomers that has a straight, rigid structure

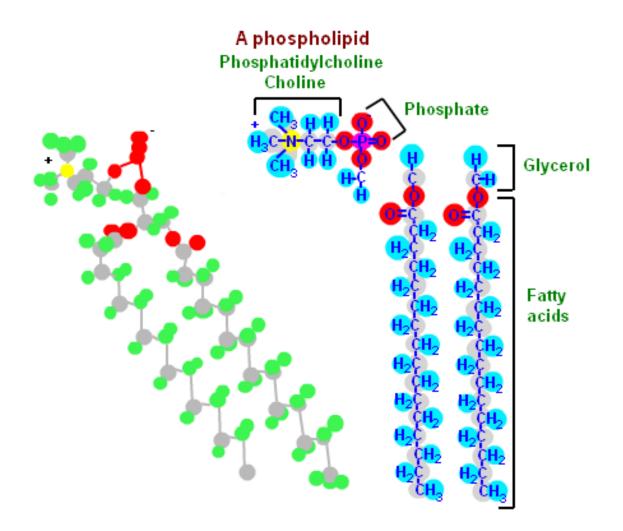
## **Functions of Carbohydrates**

- Carbohydrates can be broken down to provide energy for cells.
- Some carbohydrates are part of cell structure.
- Carbohydrates are also used for <u>short-term</u> energy storage.



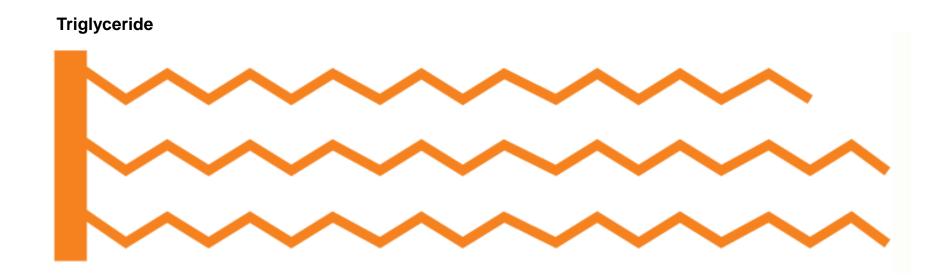
## **Structure of Lipids**

 Lipids are made of the elements carbon, hydrogen, oxygen and phosphorus.



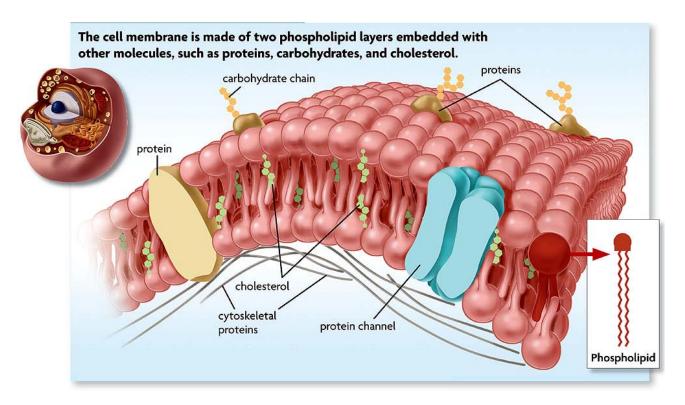
## **Structure of Lipids**

- Lipids are nonpolar molecules that include fats, oils, and cholesterol.
  - Many contain carbon chains called fatty acids.
  - Fats and oils contain fatty acids bonded to glycerol.



## Lipids have several different functions.

- broken down as a source of energy
- Store large amounts of energy long term
- make up cell membranes and other biological membranes
- used to make hormones



## **Types of Lipids**

- Fats and oils have different types of fatty acids.
  - saturated fatty acids
  - unsaturated fatty acids

#### Saturated fatty acid

Saturated fats contain fatty acids in which all carbon–carbon bonds are single bonds.

#### Unsaturated fatty acid

Unsaturated fats have fatty acids with at least one carbon—carbon double bond.

## **Examples of Lipids**

Phospholipids make up all cell membranes.

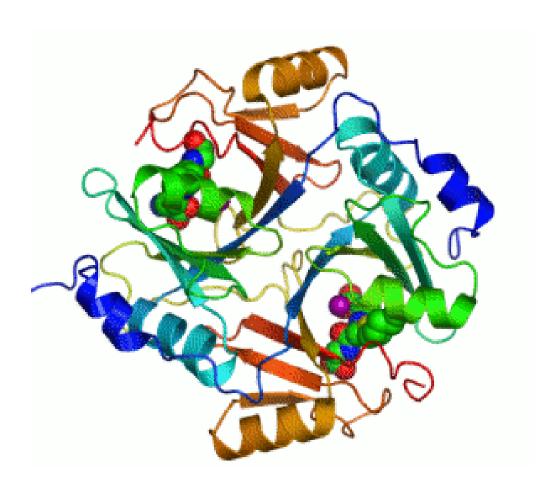
#### **Phospholipid**



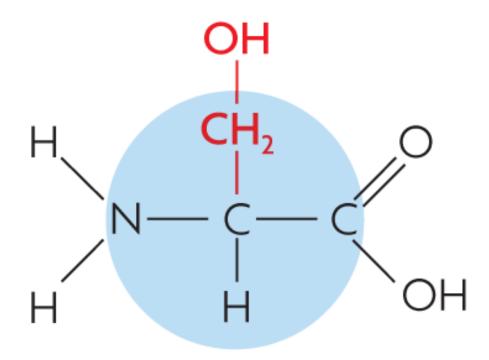
- Fats
- •Oils
- Waxes
- Cholesterol



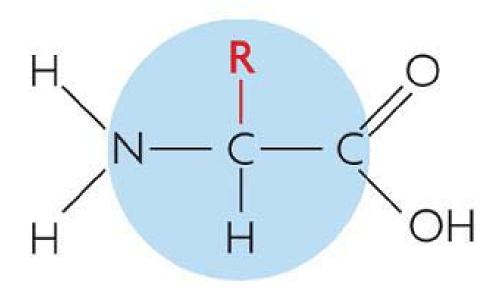
 Proteins are made of the elements carbon, hydrogen, oxygen, nitrogen and sulfur.



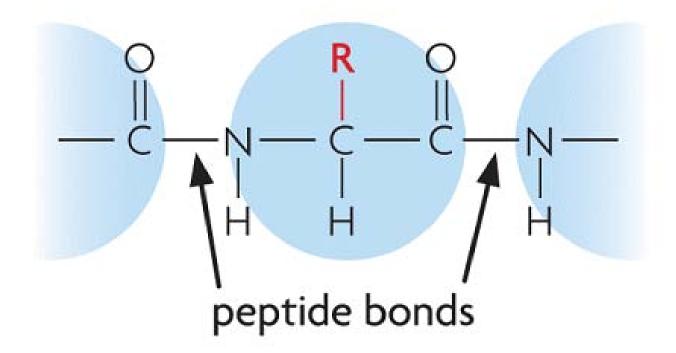
- Proteins are polymers of amino acid monomers.
  - Twenty different amino acids are used to build proteins in organisms.



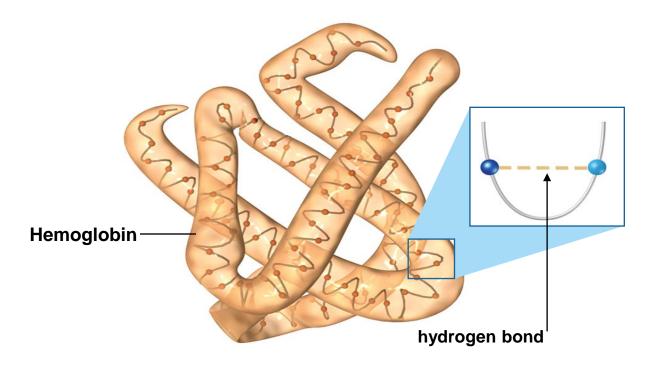
- Proteins are polymers of amino acid monomers.
  - Twenty different amino acids are used to build proteins in organisms.
  - Amino acids differ in side groups, or R groups.



- Proteins are polymers of amino acid monomers.
  - Twenty different amino acids are used to build proteins in organisms.
  - Amino acids differ in side groups, or R groups.
  - Amino acids are linked by peptide bonds.



- Proteins differ in the number and order of amino acids.
  - Amino acids interact to give a protein its shape.

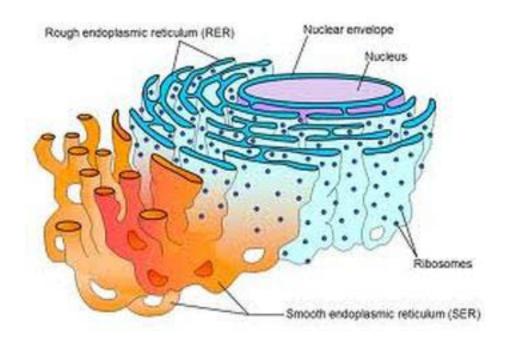


 Incorrect amino acids change a protein's structure and function.

### **Function of Proteins**

- regulate cell and body processes
- build cell structures
- building blocks of structures in organisms
- proteins called enzymes regulate chemical reactions in cells





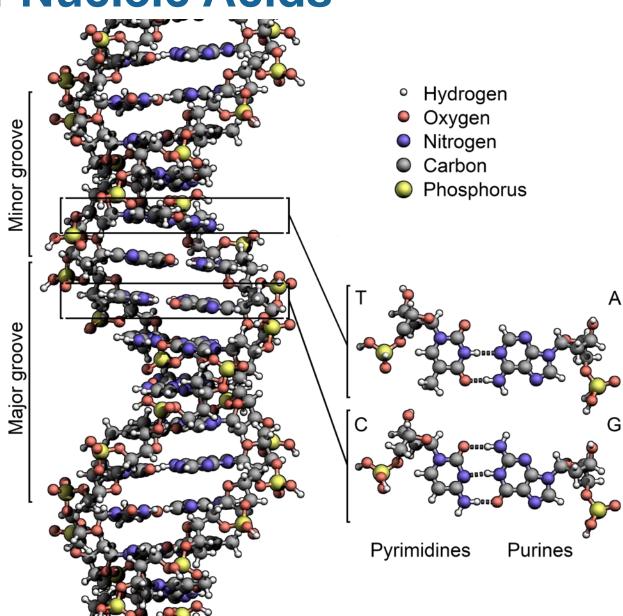
## **Examples of Proteins**

- enzymes
- skin
- hair
- connective tissues
- hemoglobin
- fibers in muscle tissue
- hormones



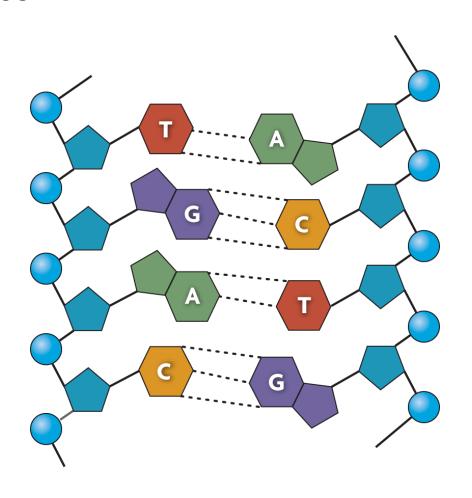
## **Structure of Nucleic Acids**

Nucleic acids are made of the elements carbon, hydrogen, oxygen, nitrogen and phosphorus.



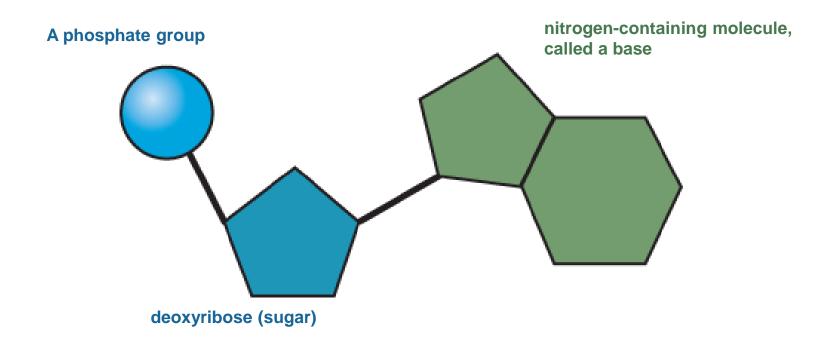
## **Structure of Nucleic Acids**

 Nucleic acids are polymers of monomers called nucleotides.



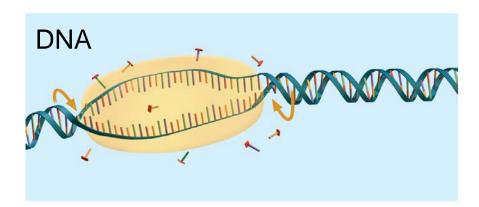
### Structure of Nucleic Acids

- Nucleic acids are polymers of monomers called nucleotides.
  - Nucleotides are made of a sugar, phosphate group, and a nitrogen base.

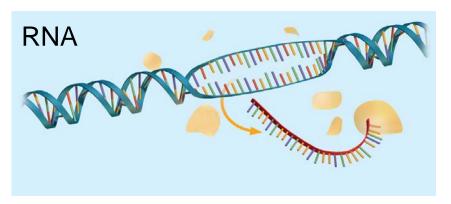


## **Types and Functions of Nucleic Acids**

DNA stores genetic information.



RNA builds proteins.



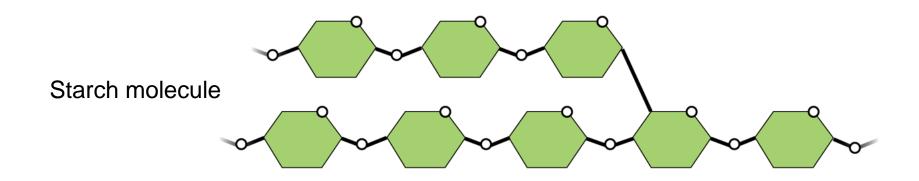
Nucleic acids are NOT used for energy!

#### **KEY CONCEPT**

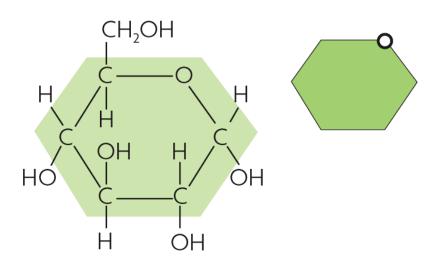
### All cells need chemical energy.



## Molecules in food store chemical energy in their bonds.



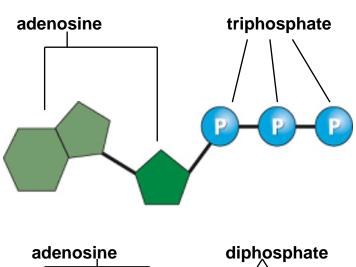
Glucose molecule



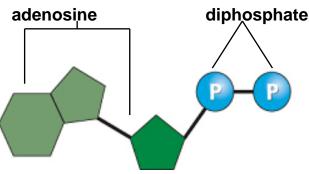
## Organisms break down carbon-based molecules to produce ATP.

- Carbohydrates are the molecules most commonly broken down to make ATP.
  - not stored in large amounts
  - up to 36 ATP from one glucose molecule

tri=3



di=2



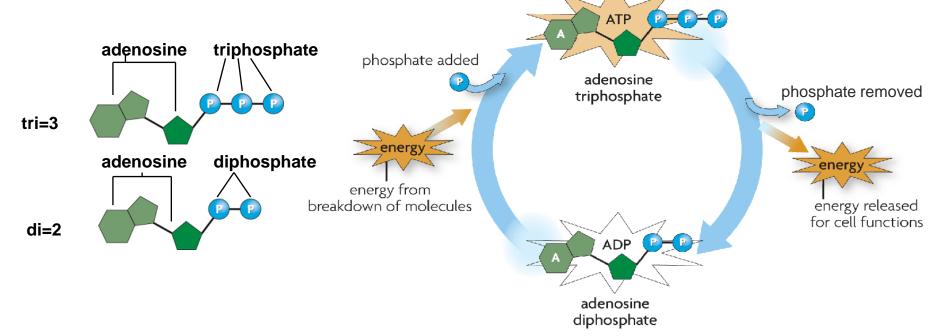
## The chemical energy used for most cell processes is carried by ATP.

 ATP transfers energy from the breakdown of food molecules to cell functions.

Energy is released when a phosphate group is removed.

ADP is changed into ATP when a phosphate group is

added.



## Fats give you the most energy <u>NOT</u> carbohydrates and proteins!

- Fats store the most energy.
  - 80 percent of the energy in your body
  - about 146 ATP from a triglyceride
- Proteins are least likely to be broken down to make ATP.
  - amino acids are not usually needed for energy
  - about the same amount of energy as a carbohydrate

MOLECULE	ENERGY
Carbohydrate	4 calories per mg
Lipid	9 calories per mg
Protein	4 calories per mg